

SYNTHESIS OF DICOUMARINYLEETHERS WITH STRUCTURES POSSIBLE FOR OREOJASMIN: COUMARINS OF *RUTA OREOJASME* FRUITS^{1,2}

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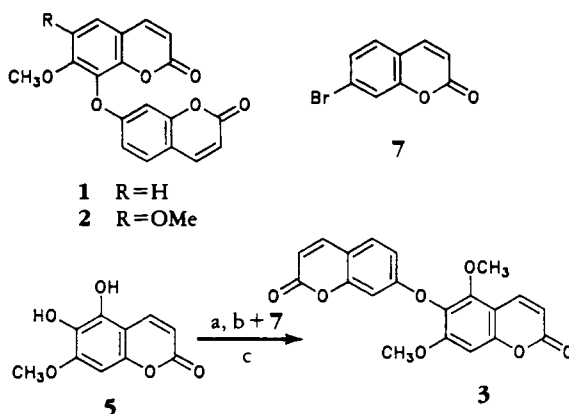
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ABSTRACT.—The structures of fatagarin [**1**] and oreojasmin [**2**], dicoumarinyl ethers isolated from the fruits of *Ruta oreojasme*, have been shown to require revision. The synthesis of 5,7-dimethoxy-7',6-oxydicoumarin [**3**] and 6,7-dimethoxy-7',5-oxydicoumarin [**4**], which are possible structures for oreojasmin, ruled out these structures for it. Gc-ms analysis of extracts of *R. oreojasme* fruits did not show the presence of dicoumarinyl ethers.

Ruta oreojasme Webb. (Rutaceae) is endemic to the Canary Islands. Apart from the simple coumarins, furanocoumarins, and pyranocoumarins isolated from various parts of the plant (1-3), two new dicoumarinyl ethers, fatagarin [**1**] and oreojasmin [**2**], have been reported from the fruits (4). In pursuing our studies on the constituents of Rutaceae, we earlier synthesized compounds with the structures proposed for

fatagarin and oreojasmin (5) and showed that these two structures required revision. In an attempt to revise the structure of oreojasmin, we have now synthesized 5,7-dimethoxy-7',6-oxydicoumarin [**3**] and 6,7-dimethoxy-7',5-oxydicoumarin [**4**], which are two possible structures for it. A gc-ms analysis of the fruits of *R. oreojasme* has also been carried out.

Because the acid hydrolysis of oreojas-

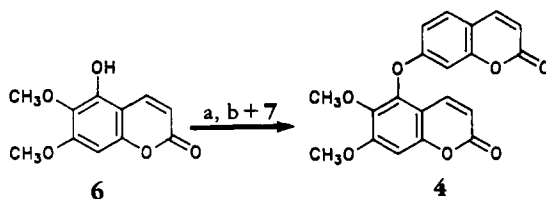


SCHEME 1. (a) NaOMe, MeOH (b) CuCl, pyridine, 120°, 18 h (c) MeI, Me₂CO, K₂CO₃, 60°, 0.5 h.

¹Part 128 in the series "Natural Product Chemistry." For part 127, see J. Reisch and M. Iding, *Monatsh. Chem.*, **120**, 363 (1989). This note is part of the Ph.D. Thesis of Anura Wickramasinghe, Münster/Peradeniya, Sri Lanka.

²In memory of the late Prof. Michael F. Grondon, The University of Ulster, Coleraine, Northern Ireland.

min [**2**] yielded 7-hydroxycoumarin and 6,7-dimethoxycoumarin (4), the synthesis of 6,7-dimethoxy-7',5-oxydicoumarin [**4**] was attempted. This is an alternative structure to the 6,7-dimethoxy-7',8-oxydicoumarin structure proposed for oreojasmin. The sodium salt of 5,6-



SCHEME 2. (a) NaOMe, MeOH (b) CuCl, pyridine, 120°, 18 h.

dihydroxy-7-methoxycoumarin [5] (6) was condensed with 7-bromocoumarin [7] under Ullmann conditions (7) and subsequently methylated. Although the formation of a mixture of 6,7-dimethoxy-7',5-oxydicoumarin [4] and 5,7-dimethoxy-7',6-oxydicoumarin [3] was expected, the only product obtained (in 20% yield) was shown by nOe experiments to be the latter.

When 6,7-dimethoxy-5-hydroxycoumarin [6] (8) was condensed with 7-bromocoumarin [7] under the same Ullmann conditions, it gave 6,7-dimethoxy-7',5-oxydicoumarin [4] in 18.5% yield. Both dicoumarinyl ethers 3, mp 201–202°, and 4, mp 204–205°, differed in their spectral characteristics from those reported (4) for oreojasmin, mp 238–239° (Table 1).

A mixture of synthetic dicoumarinyl

ethers was subjected to gc-ms analysis to select optimum conditions for their separation, and these conditions were used for the analysis of the CH₂Cl₂-soluble fraction of the MeOH extract of *R. oreojasme* fruits. Although gc-ms analysis (Table 2) showed the presence of eight coumarins, which were identified as 5,7-dimethoxycoumarin, xanthotoxin, bergapten, isopimpinellin, pimpinellin, oxypeucedanin, pabularinone, and goserol, it gave no indication that any dicoumarinyl ethers were present. Isopimpinellin was the major constituent. Because the composition of the constituents of the analyzed fruit sample is dependent on the season of collection, this analysis does not confirm the absence of dicoumarinylethers in *R. oreojasme*. However, the nonavailability of authentic fatagarin and oreojasmin

TABLE 1. ¹H-nmr Data of Oreojasmin [2] and Synthetic Dicoumarinylethers 3 and 4.

Proton	Compound		
	2 ^a	Oxydicoumarin 3	Oxydicoumarin 4
H-3, -3'	6.34	6.30	6.26
(d, J = 10 Hz)			6.32
H-4, -4'	7.48	7.67	7.67
(d, J = 10 Hz)	7.67	7.96	7.71
OMe	3.93	3.84	3.75
	3.97	3.97	3.99
H-5'	7.32	7.43	7.44
(d, J = 8.5 Hz)			
H-6'	↑	6.88	↑
		(dd, J = 8.5, 2.4 Hz)	
H-8'	multiplet	6.77	multiplet
	6.80–7.10	(d, J = 2.4 Hz)	6.82–6.88
H-8	↓	6.72	↓
		(s)	

^aData for oreojasmin [2] are taken from González *et al.* (4).

TABLE 2. Gc-ms Analysis of the CH₂Cl₂ Extract of *Ruta oreojasme* Fruits.

Constituent	Gc retention time (min:sec)	(100%) <i>m/z</i>	<i>m/z</i>
5,7-Dimethoxycoumarin	28:27	206	206
Xanthotoxin	29:25	216	216
Bergapten	29:54	216	216
Isopimpinellin	32:42	231	246
Pimpinellin	34:28	59	246
Oxypeucedanonin	36:16	59	286
Pabularinone	36:41	43	286
Gosferol	37:56	202	286

prevents us from carrying out further experiments in order to elucidate their structures.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—

Melting points were determined on a Kofler hot stage apparatus and are uncorrected. Ir spectra were recorded for KBr discs on a Pye-Unicam SP 3-200 and uv spectra in MeOH solutions on a Carl Zeiss DMR 21 spectrophotometer. ¹H- and ¹³C-nmr spectra were measured at Varian Gemini 200 MHz in CDCl₃ solutions using TMS as internal standard. Mass spectral analyses were carried out at 70 eV on a MAT 44 S spectrometer.

Gc-ms analyses were recorded on a Varian MAT CH 7A, 70 eV mass spectrometer connected to a Shimadzu GC-8AX gas chromatograph fitted with a quartz capillary column (50 m, 0.32 mm i.d.) coated with HP-1. Temperature programming was from 70° to 300° at 6°/min, injector temperature 300°, split valve 1:10. Merck Si gel 60 F₂₅₄ coated on aluminum sheets and glass plates was used for analytical and preparative (2 mm) chromatography.

The dried, ground plant material (5 g) was extracted with 95% MeOH for three successive 24-h periods at 20°, and the concentrated extract was partitioned between CH₂Cl₂ and H₂O. The CH₂Cl₂ layer was concentrated to give the CH₂Cl₂ extract (53 mg).

5,7-DIMETHOXY-7',6-OXYDICOUMARIN [3].—7-Bromocoumarin [7] (225 mg, 1 mmol) and the sodium salt of 5,6-dihydroxy-7-methoxycoumarin [5] (231 mg, 1 mmol) were refluxed in pyridine (5 ml) with CuCl (18 mg, 0.18 mmol) for 18 h under N₂. Acidification (2 N HCl) followed by extraction with hexane and then with CH₂Cl₂ gave two extracts. On chromatographic separation, the hexane extract gave back 7-bromocoumarin (95 mg).

The CH₂Cl₂ extract was concentrated in vacuo and refluxed in dry Me₂CO (10 ml) with MeI (300 mg) and anhydrous K₂CO₃ (250 mg) for 0.5 h.

The product on chromatographic separation gave 5,7-dimethoxy-7',6-oxydicoumarin [3] (43 mg, 20%) (Scheme 1) as colorless crystals from MeOH: mp 201–202°; found C 65.58, H 3.88, C₂₀H₁₄O₇ requires C 65.56, H 3.85%; ir ν max cm⁻¹ 1730 (C=O), 1600 (C=C, arom.), 1250, 1120 (C=O), 1100, 840 (C-H, arom.); uv λ max nm (log ϵ) 207 (4.64), 228 sh (4.30), 252 sh (3.84), 294 sh (4.31), 321 (4.51); ¹H nmr see Table 1; ¹³C nmr δ (ppm) 56.8, 62.2 (2 OMe), 96.2 (C-8'), 103.4 (C-6'), 107.7 (C-4a), 112.7 (C-4'a), 112.6 113.5, and 114.5 (C-8, -3', -3), 129.5 (C-5'), 131.5 (C-6), 138.9 (C-4), 143.5 (C-4'), 150.0 (C-7), 153.7 (C-5), 156.2 and 156.9 (C-8a, -8'a), 161.1, 161.2, and 161.4 (C-2, -2', -7'); ms *m/z* (%) [M]⁺ 366 (85), [M-CO]⁺ 338 (10), 221 (26), 193 (26), [C₇H₅]⁺ 89 (100).

6,7-DIMETHOXY-7',5-OXYDICOUMARIN [4].—7-Bromocoumarin [7] (270 mg, 1.2 mmol), the sodium salt of 6,7-dimethoxy-5-hydroxycoumarin [6] (147 mg, 0.6 mmol), and CuCl (12 mg, 0.12 mmol) in pyridine (2 ml), when reacted by the above method, gave recovered 7-bromocoumarin (190 mg) and 6,7-dimethoxy-7',5-oxydicoumarin [4] (24 mg, 18.5%) (Scheme 2) as colorless crystals from MeOH: mp 204–205°; found C 65.44, H 3.92, C₂₀H₁₄O₇ requires C 65.56, H 3.85%, ir ν max cm⁻¹ 1725 (C=O), 1610 (C=C, arom.), 1450, 1260, 1130 (C-O), 830 (C-H, arom.); uv λ max nm (log ϵ) 207 (4.67), 225 sh (4.45), 291 sh (4.33), 320 (4.52); ¹H nmr see Table 1; ¹³C nmr δ (ppm) 56.7 and 61.5 (2-OMe), 98.5 (C-8'), 104.0 (C-8), 107.6 (C-4a), 112.7, 114.3, and 114.9 (C-3, -3', -6'), 114.7 (C-4'a), 129.6 (C-5'), 138.0 (C-4'), 138.1 (C-6), 143.2 (C-7), 143.3 (C-4), 152.1 (C-5), 156.1 and 157.7 (C-8a, -8'a), 160.9 and 161.4 (C-2, -2', -7'); ms *m/z* (%) [M]⁺ 366 (100), [M-Me]⁺ 351 (35), [M-Me-CO]⁺ 323 (28), [C₇H₅]⁺ 89 (95).

ANALYSIS OF FRUIT EXTRACT.—The CH₂Cl₂ extract of *R. oreojasme* fruits was subjected to gc-ms analysis. The results are shown in Table 2.

The spectra of the compounds were compared with those of authentic specimens and with published data (9, 10).

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